

Published on Web 05/12/2009

Nickel-Catalyzed Decarboxylative Carboamination of Alkynes with Isatoic Anhydrides

Yasufumi Yoshino, Takuya Kurahashi,* and Seijiro Matsubara*

Department of Material Chemistry, Graduate School of Engineering, Kyoto University, Kyoto 615-8510, Japan

Received February 2, 2009; E-mail: tkuraha@orgrxn.mbox.media.kyoto-u.ac.jp; matsubar@orgrxn.mbox.media.kyoto-u.ac.jp

Addition of C and N fragments across unsaturated C–C bonds to form C–C and C–N bonds simultaneously (i.e., carboamination) would be a potentially useful and straightforward method for introducing a nitrogen atom function into organic molecules.^{1–4} Recently, we demonstrated that oxidative addition of an amide to a Ni(0) complex and subsequent decarbonylation afford C–Ni–N species,^{5,6} which lead to the carboamination of alkynes (Scheme 1a).⁷ A key step in the reaction is migratory extrusion of CO. The

 $\ensuremath{\textit{Scheme 1.}}$ Nickel-Catalyzed Carboamination via Decarbonylation and Decarboxylation



extruded CO should be removed from the reaction system for an efficient catalytic cycle, since it interacts very strongly with Ni and saturates the coordination sites, thus interfering with regeneration of an active catalyst or causing the reverse. We postulated that an active species (C-Ni-N) might be generated alternatively via oxidative addition of a carbamate to Ni(0) followed by decarboxylation (Scheme 1b). This would allow for robust carboamination of alkynes, since CO₂ coordinates to Ni much less strongly than CO does.^{8,9} Thus, we attempted the addition of isatoic anhydride **1** to alkyne **2** to form quinolone **3** using a Ni(0) catalyst.^{7,10}

Initially, *N*-phenylisatoic anhydride (**1a**) was treated with 4-octyne (**2a**), 5 mol % Ni(cod)₂, and 5 mol % PCy₃ in toluene at 80 °C for 24 h. This led to formation of quinolone **3aa** in 97% isolated yield (Table 1, entry 1). Under the same conditions, the decar-

 $\mbox{\it Table 1.}$ Nickel-Catalyzed Decarboxylative Carboamination of 4-Octyne with Isatoic Anhydrides^a

R ²	$ \begin{array}{c} 0 \\ 1 \\ R^1 \end{array} $	Pr .5 equiv. 2a	Ni(cod) ₂ (PCy ₃ (5 m toluene, 8	5 mol%) nol%) 0 °C, 24 h	R ²	0 Pr N Pr 3 R ¹
entry	R ¹	R ²	R ³	1	3	yield (%) ^b
1	Ph	Н	Н	1a	3aa	97
2	4-MeO-C ₆ H ₄	Н	Н	1b	3ba	95
3	$4-CF_3-C_6H_4$	Н	Н	1c	3ca	96
4	Me	Н	Н	1d	3da	93
5	cyclopropyl	Н	Н	1e	3ea	69 ^c
6	Boc	Н	Н	1f	3fa	99 ^c
7	Ph	MeO	Н	1g	3ga	95
8	Ph	Η	CF_3	1h	3ha	92

^{*a*} Reactions were carried out using Ni(cod)₂ (5 mol %), PCy₃ (5 mol %), isatoic anhydride (0.5 mmol), and **2a** (0.75 mmol) in 2 mL of toluene at 80 °C for 24 h. ^{*b*} Isolated yields. ^{*c*} The reaction was carried out at 60 °C to prevent decomposition of the substrate.

boxylative carboaminations of **2a** with *N*-arylisatoic anhydrides possessing electron-donating and -withdrawing groups also provided quinolones in 95 and 96% yield, respectively (entries 2 and 3). The reaction of *N*-methylisatoic anhydride (**1d**) with **2a** also gave the product **3da** in 93% yield (entry 4). The reactions of *N*cyclopropyl-substituted and *N*-Boc-protected isatoic anhydrides with **2a** also afforded the correspondingly substituted quinolones in 69 and 99% yield, respectively (entries 5 and 6). Ring-substituted *N*-phenylisatoic anhydrides possessing electron-donating and -withdrawing groups also provided the corresponding quinolones in excellent yields (entries 7 and 8).

We next investigated the scope of the reaction using various alkynes. The carboamination was also compatible with an aryl-substituted alkyne (Table 2, entry 1), affording the corresponding quinolone 3ab in 66% yield. The reaction of 1a with the

Table 2. Nickel-Catalyzed Decarboxylative Carboamination of Alkynes with $\mathbf{1a}^a$



^{*a*} Reactions were carried out using Ni(cod)₂ (5 mol %), PCy₃ (5 mol %), **1a** (0.5 mmol), and alkyne (0.75 mmol) in 2 mL of toluene at 80 °C for 24 h. ^{*b*} Isolated yields. ^{*c*} Pyridine was used as the solvent instead of toluene. ^{*d*} Ratio of regioisomers. ^{*e*} The structure of the major product was determined by single-crystal X-ray structure analysis. ^{*f*} The structure of the major product was determined by NOE analysis.

unsymmetrical alkynes 2c-e gave the quinolones consisting of regioisomers in a range of ratios from 2/1 to 6/1 (entries 2–4), whereas reaction with 2f gave quinolone 3af with high selectivity (entry 5). Bulky trimethylsilyl-substituted alkynes 2g and 2h also reacted with 1a to provide adducts with complete regiocontrol in 90 and 94% yield, respectively (entries 6 and 7). Trimethylsilyl-substituted enyne 2i also reacted with 1a to furnish 3ai regiose-lectively in 69% yield.

A reaction pathway based on the observed results that accounts for the formation of quinolones **3** is outlined in Scheme 2. In view

Scheme 2. Plausible Pathway for Decarboxylative Carboamination of Alkynes with Isatoic Anhydrides



of the mechanism of the previously reported nickel-catalyzed crosscoupling reaction of anhydrides with organometallic reagents, it is reasonable to consider that the catalytic cycle of the present reaction should consist of oxidative addition of an anhydride O-CO bond to nickel.¹¹ Subsequent decarboxylation and coordination of alkyne affords Ni(II) intermediate 6. The alkyne then inserts into the acyl-nickel bond to give nickelacycle 7, which undergoes reductive elimination to give 3 and regenerate the starting Ni(0). The regioselectivity of the reaction can be rationalized in terms of the direction of alkyne insertion, in which the repulsive steric interaction is minimal between the bulkier R^L and the PCy₃ ligand on the nickel to give nickel(II) intermediate 6a. Thus, the reaction of 1a with 2e using less sterically hindered ligands, such as PPh₃ and PMe₃, affords 3ae with lower regioselectivity (Table 3, entries 2 and 3), while the reaction using a more sterically hindered N-heterocarbene ligand shows higher regioselectivity (entry 4).¹²

Table 3. Effects of the Ligand in Nickel-Catalyzed Decarboxylative Carboamination of 2e with $1a^{\rm a}$

اللہ ا	$ \begin{array}{c} 0 \\ \hline N \\ Ph \\ 2e \end{array} $	Ni(cod) ₂ (5 mol%) ligand (5 mol%) toluene, 80 °C, 24 h	O N N N N N N N N N N N N N N N N N N N
entry	ligand	ratio of regioisomers	yield (%) ^b
1	PCy ₃	6/1	94
2	PPh ₃	3/1	94
3	PMe ₃	3/2	85
4	$\rm NHC^{c}$	10/1	50
5	P'Bu ₃	3/1	94
6	$Cy_2PCH_2CH_2PCy_2$	5/1	74

^{*a*} Reactions were carried out using Ni(cod)₂ (5 mol %), ligand (5 mol %), **1a** (0.5 mmol), and **2a** (0.75 mmol) in 2 mL of toluene at 80 °C for 24 h. ^{*b*} Isolated yields. ^{*c*} 1,3-Bis(2,6-diisopropyl)imidazol-2-ylidene.

In summary, we have demonstrated a new nickel-catalyzed reaction of alkynes with isatoic anhydrides to provide quinolones. We have shown that carbamates are susceptible to oxidative addition of a Ni(0) complex, which allows intermolecular addition to alkynes via decarboxylation. Further efforts to expand the scope of the chemistry and studies of the detailed mechanism are currently underway in our laboratories.

Acknowledgment. This work was supported by Grants-in-Aid from the Ministry of Education, Culture, Sports, Science, and Technology, Japan. T.K. also acknowledges the Novartis Foundation (Japan) for the Promotion of Science and the Takeda Pharmaceutical Company Award in Synthetic Organic Chemistry, Japan.

Supporting Information Available: Experimental procedures, spectroscopic and analytical data for new compounds, and CIF files for **3ac**, **3ad**, and **3af–ai**. This material is available free of charge via the Internet at http://pubs.acs.org.

References

- For carbonylation of azides and azetidenes, see: (a) Calet, S.; Urso, F.; Alper, H. J. Am. Chem. Soc. **1989**, 111, 931. (b) Roberto, D.; Alper, H. J. Am. Chem. Soc. **1989**, 111, 7539. For carbonylation of allylamines, see: (c) Murahashi, S.-I.; Imada, Y.; Nishimura, K. Tetrahedron **1991**, 50, 453.
- (2) For PtCl₂-catalyzed intramolecular carboamination, see: (a) Shimada, T.; Nakamura, I.; Yamamoto, Y. J. Am. Chem. Soc. 2004, 126, 10546. (b) Fürstner, A.; Davies, P. W. J. Am. Chem. Soc. 2005, 127, 15024.
- (3) Ruck, R. T.; Zuckerman, R. L.; Krska, S. W.; Bergman, R. G. Angew. Chem., Int. Ed. 2004, 43, 5372.
- (4) (a) Liu, Z.; Larock, R. C. J. Am. Chem. Soc. 2005, 127, 13112. (b) Yoshida, H.; Shirakawa, E.; Honda, Y.; Hiyama, T. Angew. Chem., Int. Ed. 2002, 41, 3247.
- (5) For catalytic reactions involving C-N cleavage, see: (a) Lei, Y.; Wrobleski, A. D.; Golden, J. E.; Powell, D. R.; Aubé, J. J. Am. Chem. Soc. 2005, 127, 4552. (b) Ito, M.; Sakaguchi, A.; Kobayashi, C.; Ikariya, T. J. Am. Chem. Soc. 2007, 129, 290. (c) Ueno, S.; Chatani, N.; Kakiuchi, F. J. Am. Chem. Soc. 2007, 129, 6098. (d) Miura, T.; Yamauchi, M.; Murakami, M. Org. Lett. 2008, 10, 3085.
- (6) For pioneering works on C-N activation with transition-metal complexes, see: (a) Chan, Y. W.; Renner, M. W.; Balch, A. L. Organometallics 1983, 2, 1888. (b) Bonanno, J. B.; Henry, T. P.; Neithamer, D. R.; Wolczanski, P. T.; Lobkovsky, E. B. J. Am. Chem. Soc. 1996, 118, 5132. (c) Tayebani, M.; Gambarotta, S.; Yap, G. Organometallics 1998, 17, 3639.
- (7) Ni-catalyzed decarbonylative addition of phthalimides to alkynes provides isoquinolones. For details, see: Kajita, Y.; Kurahashi, T.; Matsubara, S. J. Am. Chem. Soc. 2008, 130, 6058.
- (8) The binding energy of Ni(CO₂) is estimated to be 18 kcal/mol, while that of Ni(CO) is estimated to be 30 kcal/mol. For details, see: (a) Blomberg, M. R. A.; Brandemak, U. B.; Siegbahn, P. E. M.; Wennerberg, J.; Bauschlicher, C. W. J. Am. Chem. Soc. **1988**, 110, 6650. (b) Galan, F.; Fouassier, M.; Tranquille, M.; Mascetti, J.; Papai, I. J. Phys. Chem. A **1997**, 101, 2626.
- (9) For pioneering works on decarboxylative insertion reactions of α,βunsaturated carbonyl compounds, see: (a) Inoue, Y.; Ajioka, S.; Toyofuku, M.; Mori, A.; Fukui, T.; Kawashima, Y.; Miyano, S.; Hashimoto, H. J. Mol. Catal. **1985**, 32, 91. (b) Wang, C.; Tunge, J. A. Org. Lett. **2005**, 7, 2137. (c) Wang, C.; Tunge, J. A. Org. Lett. **2006**, 8, 3211. (d) Shintani, R.; Park, S.; Hayashi, T. J. Am. Chem. Soc. **2007**, 129, 14866.
- (10) Quinolone represents an important class of naturally occurring alkaloids that display a wide range of biological activities and are useful intermediates in the synthesis of a variety of hetero- and carbocyclic compounds. For recent reviews, see: (a) Mitscher, L. A. *Chem. Rev* 2005, *105*, 559. (b) Emami, S.; Shafiee, A.; Foroumadi, A. *Mini-Rev. Med. Chem.* 2006, *6*, 375.
- (11) (a) Trost, B. M.; Chen, F. *Tetrahedron Lett.* **1971**, *12*, 2603. (b) Sano, K.; Yamamoto, T.; Yamamoto, A. *Chem. Lett.* **1984**, *13*, 941. (c) Fischer, R.; Walther, D.; Kempe, R.; Sieler, J.; Schönecker, B. J. Organomet. Chem. **1993**, *447*, 131. (d) O'Brien, E. M.; Bercot, E. A.; Rovis, T. J. Am. Chem. Soc. **2003**, *125*, 10498. (e) Johnson, J. B.; Bercot, E. A.; Rowley, J. M.; Coates, G. W.; Rovis, T. J. Am. Chem. Soc. **2007**, *129*, 2718.
- (12) These results may rule out the possibility that the reaction mechanism involves formation of iminoketene and subsequent cycloaddition with alkyne. For iminoketene cycloaddition, see: Kametani, T.; Higa, T.; Loc, C. V.; Ihara, M.; Koizumi, M.; Fukumoto, K. J. Am. Chem. Soc. 1976, 98, 6186.

JA900805Y